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The effect of additives on gibbsite auto-precipitation and bauxite residue flocculation when processing goethitic bauxites

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Introduction

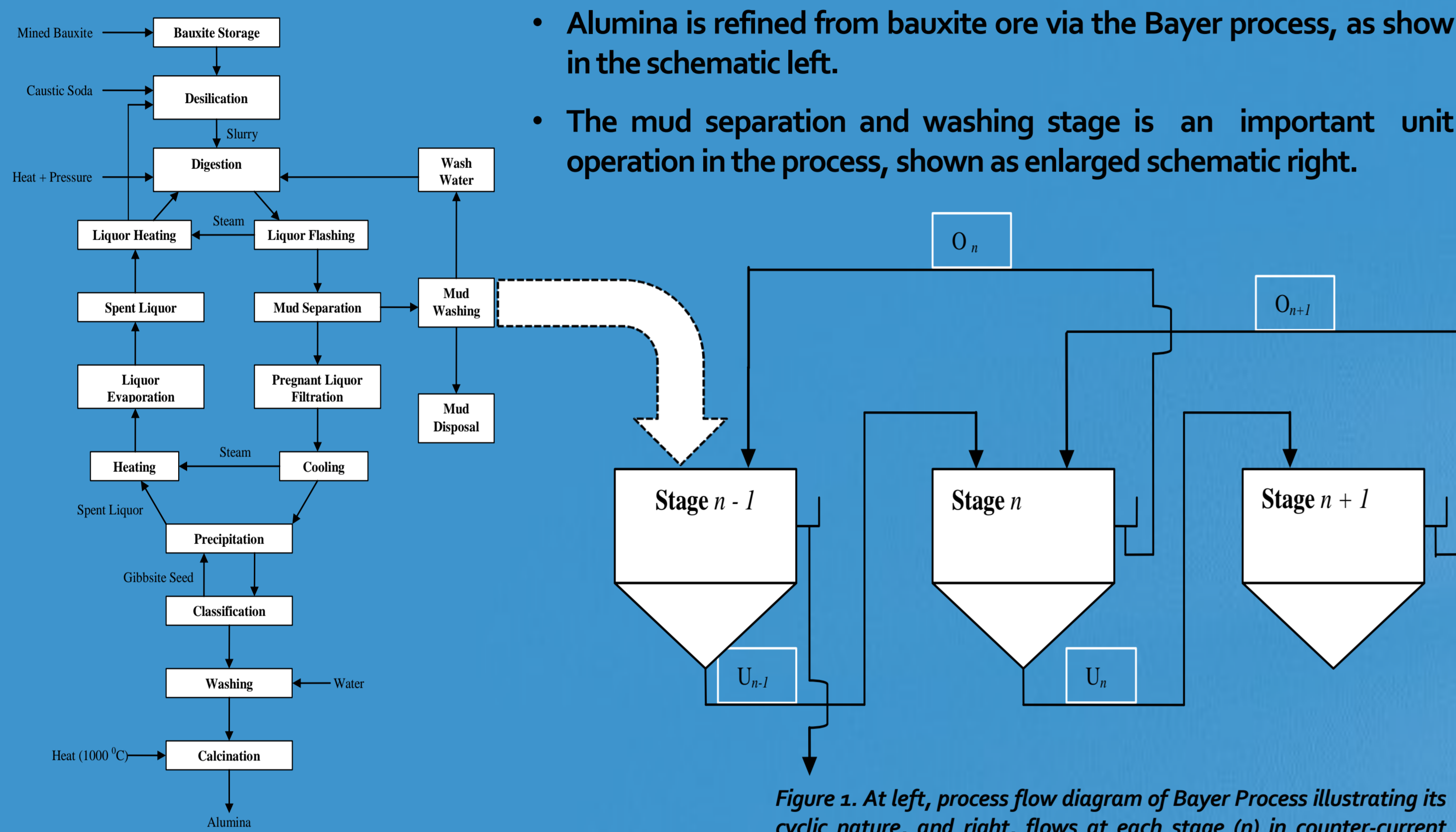


Figure 1. At left, process flow diagram of Bayer Process illustrating its cyclic nature, and right, flows at each stage (n) in counter-current decantation (O = overflow; U = underflow).

Motivation

- Alumina producers often experience premature alumina crystallisation (gibbsite auto-precipitation) in the mud separation and washing stage: $Al(OH)_4^- \rightleftharpoons Al(OH)_3 + OH^-$
- Recent changes to bauxite ore quality at a Jamaican Bayer refinery saw major losses of alumina and caustic from the process, attributable to certain mineral phases in the bauxite.

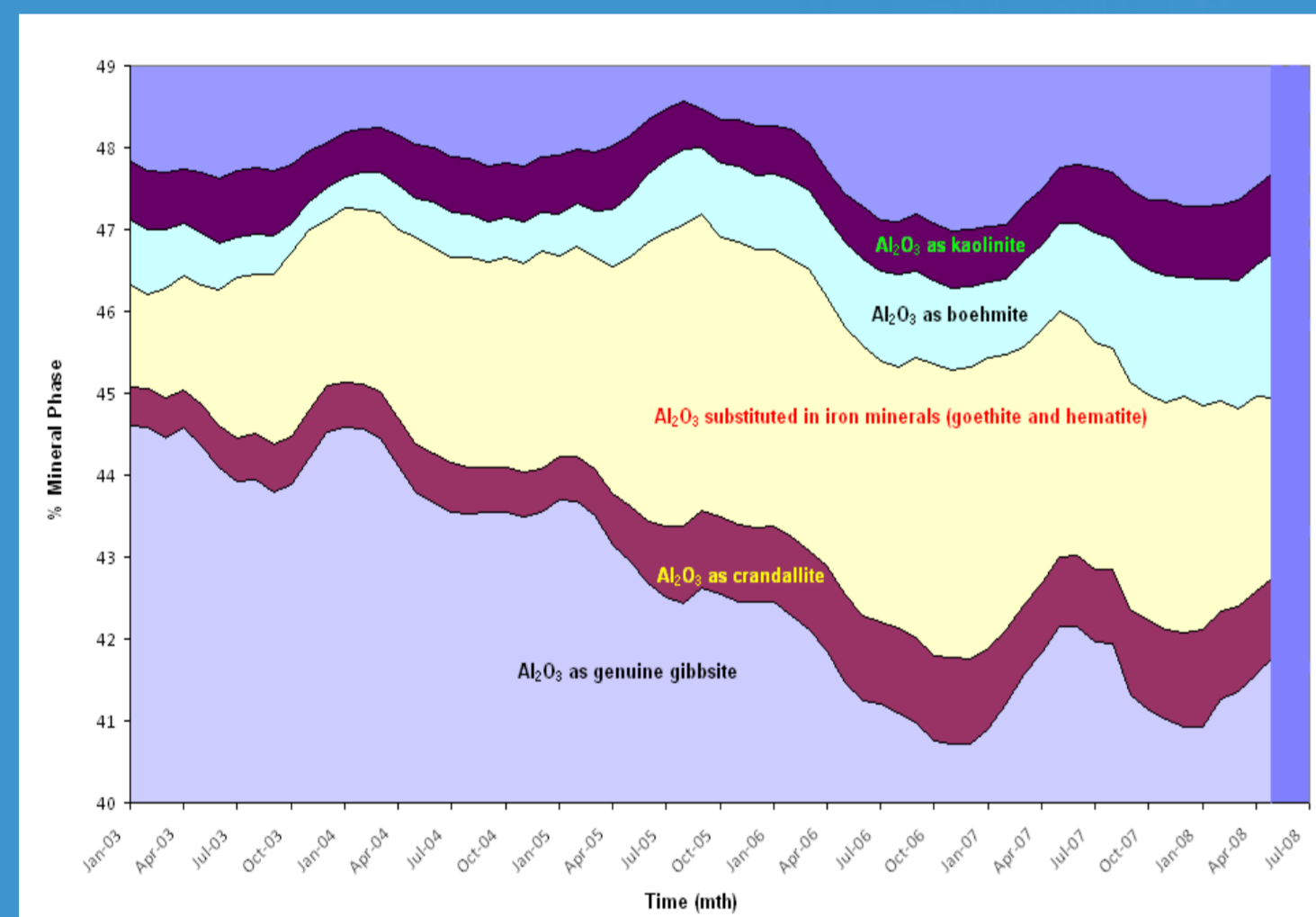


Figure 2. Time variance of bauxite ore quality at Jamaican mine³.

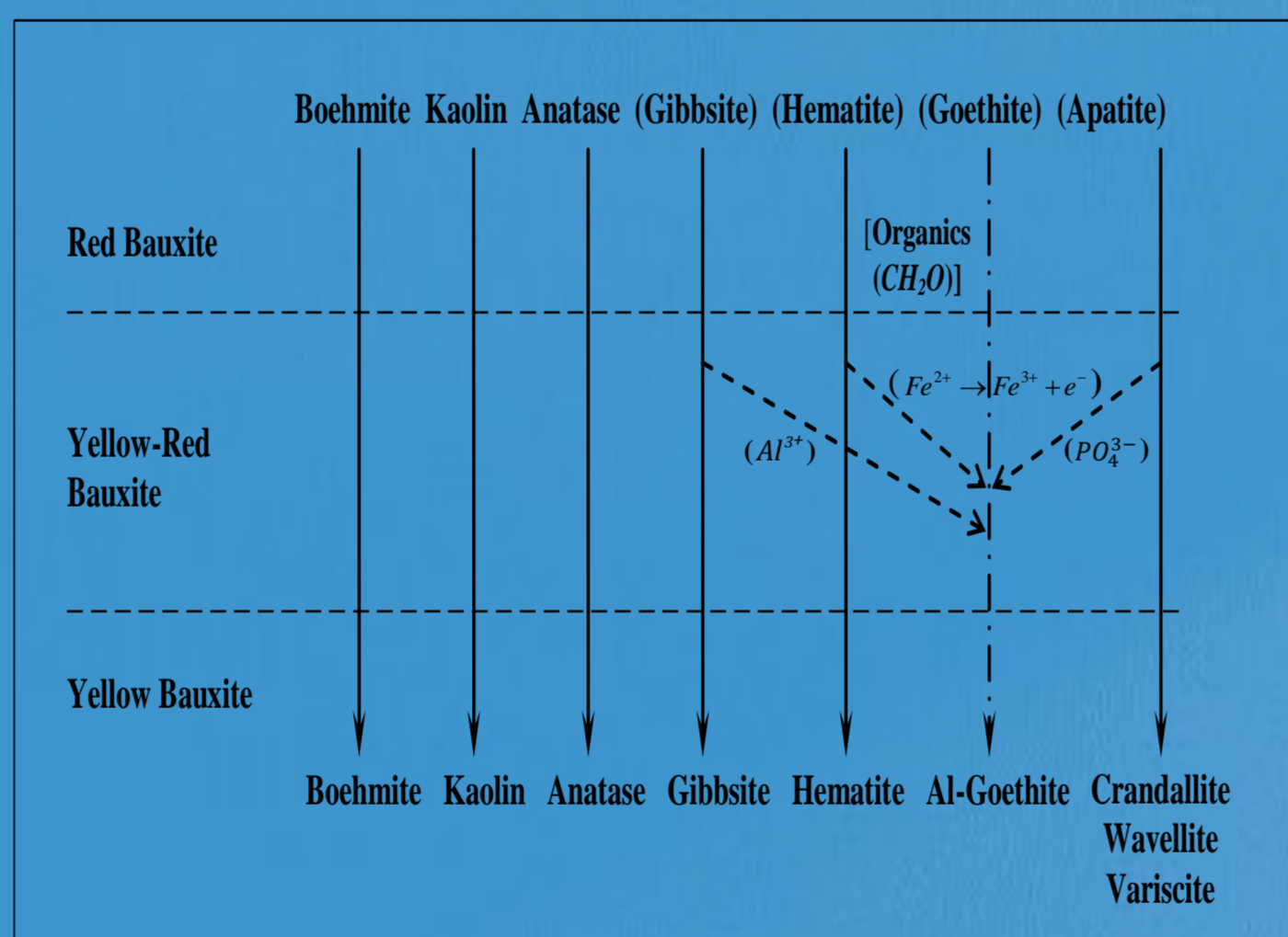


Figure 3. Mineral transformations of bauxite ores due to weathering

Experimental

Table 1. Bayer process terminologies and concentrations, and their molar equivalents.

Bayer Process Terminology	Bayer Process Concentration (g/L ³)	Molar Equivalent Terminology	Molar Equivalent Concentration (mol L ⁻³)
A as Al ₂ O ₃	156	NaAl(OH) ₄	3.06
CaS Na ₂ CO ₃	240	NaOH	1.47
		NaAl(OH) ₄	3.06
TTS as Na ₂ CO ₃	300	Na ₂ CO ₃	5.66
Carbonate as Na ₂ CO ₃	60	Na ₂ CO ₃	0.57

- Bauxite ores were digested at 150 °C for 30 minutes (A/C ≈ 0.65) to give bauxite residue seeds used in the study.
- The effect of additives on gibbsite auto-precipitation was monitored at 75 °C.
- Gibbsite crystallisation and bauxite residue flocculation were monitored by FBRM.



Figure 4. Baskerville pilot autoclave used to digest bauxite ores.

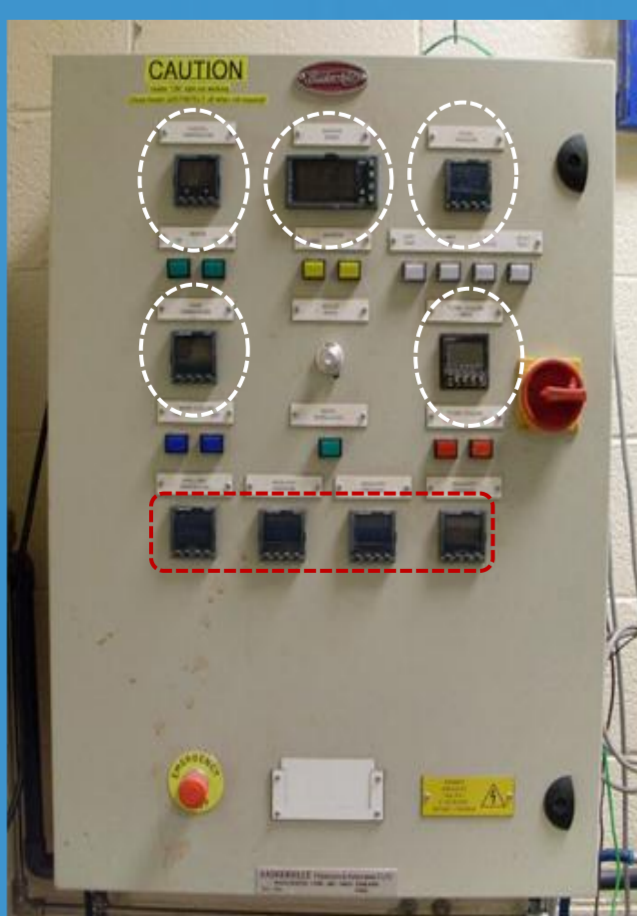


Figure 5. Baskerville Eurotherm process control unit.



Figure 6. Parr 4560 benchtop autoclave used for digestion.

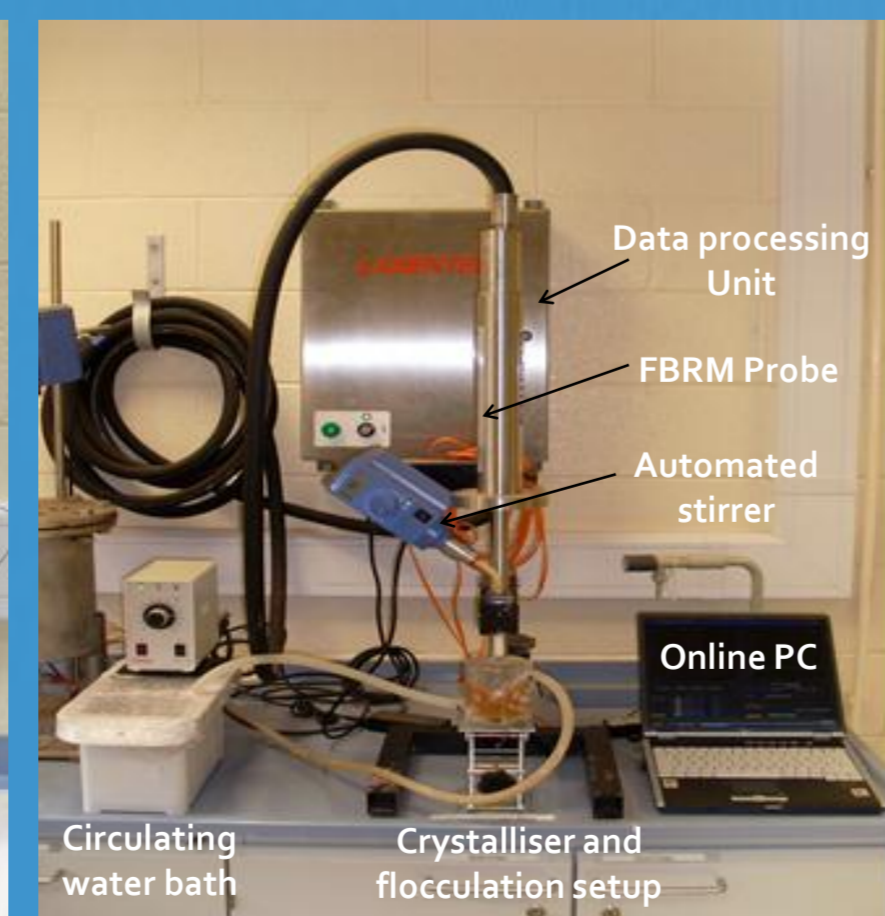


Figure 7. Experimental setup for flocculation and gibbsite crystallisation studies.

Results & Discussion

Gibbsite auto-precipitation experiments

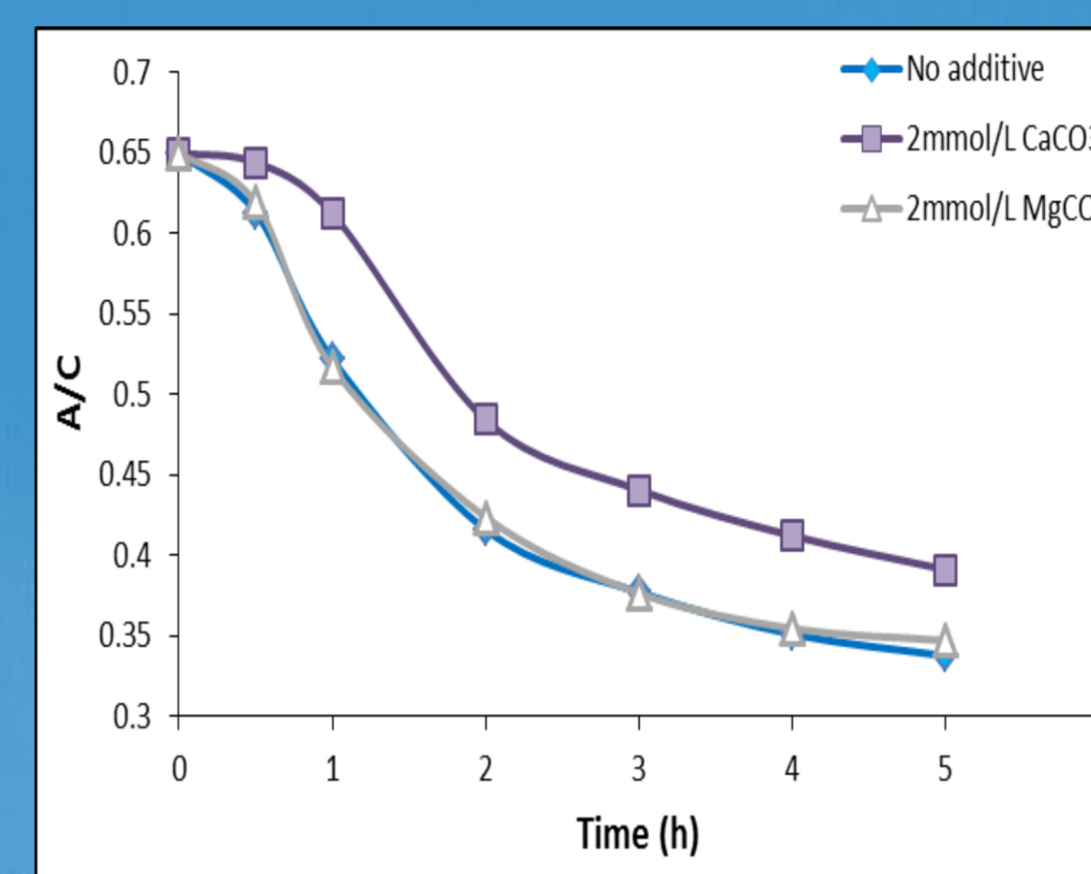


Figure 8. Gibbsite auto-precipitation with no additive and 100 ppm HX800.

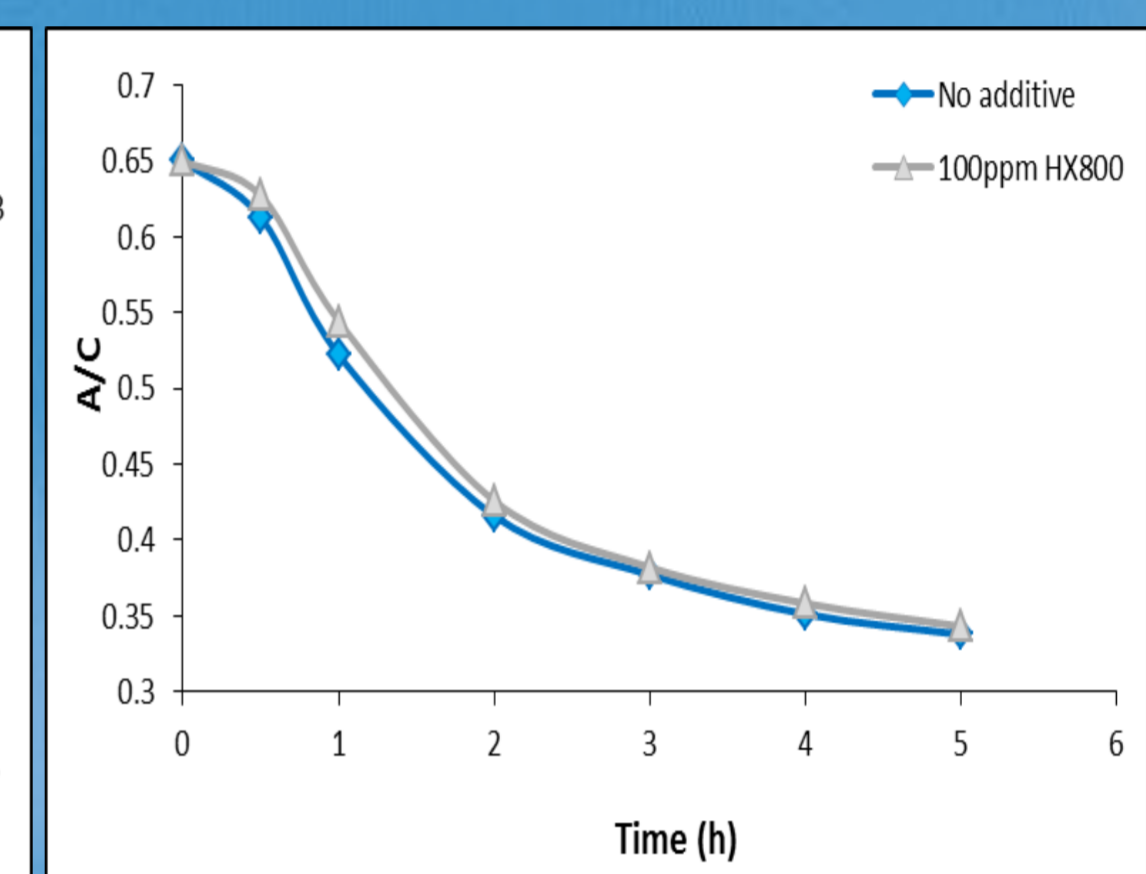


Figure 9. Gibbsite auto-precipitation with no additive, 2 mmol L⁻¹ CaCO₃ and MgCO₃.

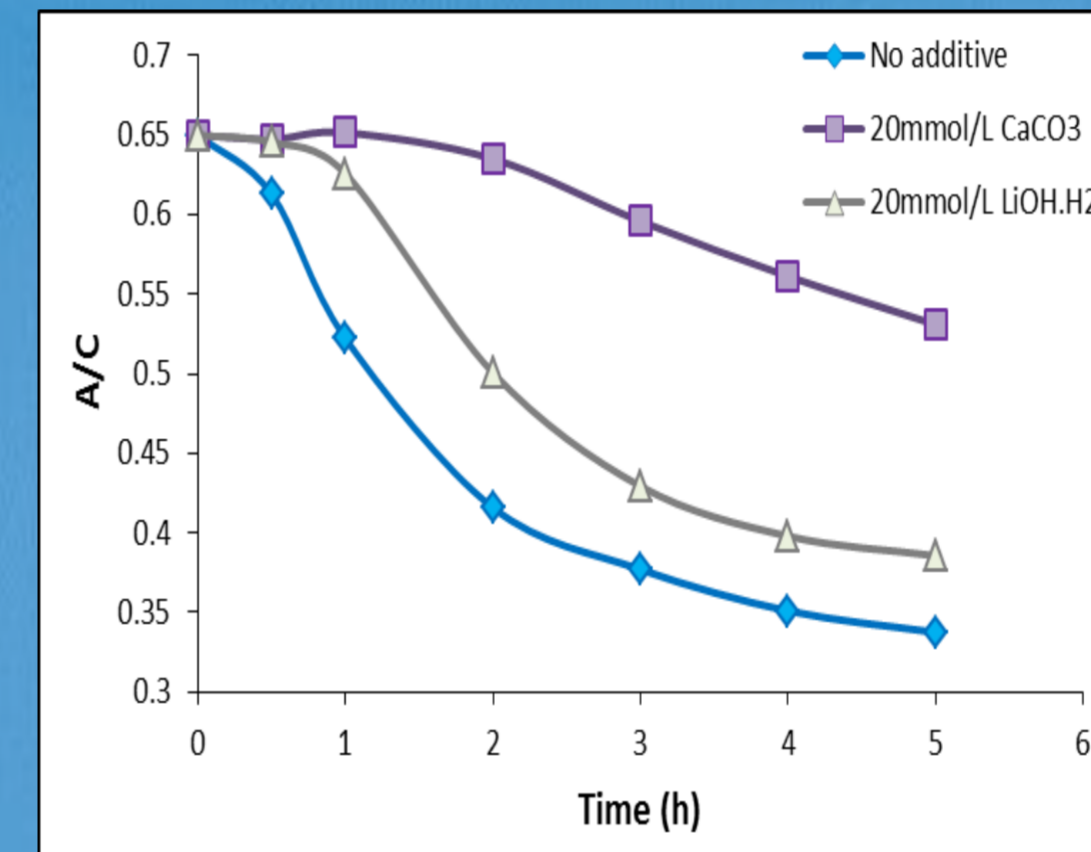


Figure 10. Gibbsite auto-precipitation with no additive, 20 mmol L⁻¹ CaCO₃ and LiOH.H₂O.

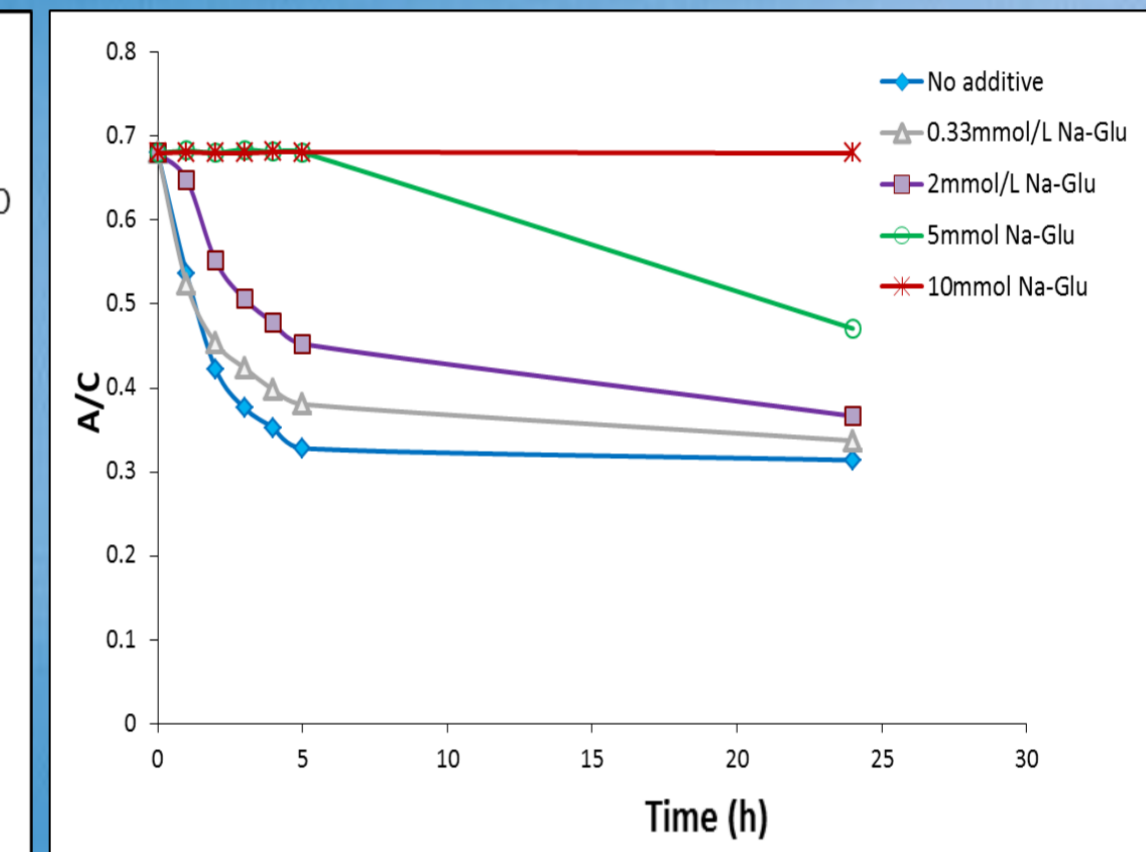
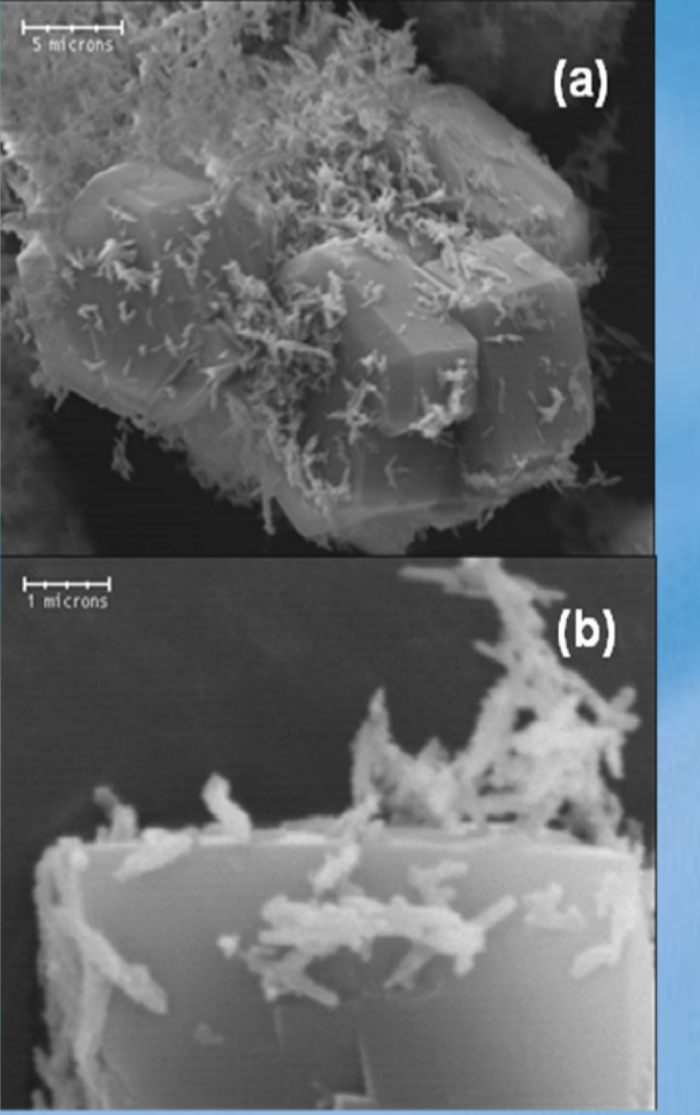


Figure 11. Gibbsite auto-precipitation with additive and a range of gluconate doses.

Figure 12. (a) Far view; (b) close view. Gibbsite auto-precipitation with goethite seed present (fine needle-like crystals).



- Order of effect: HX800 ≈ MgCO₃ < LiOH.H₂O < CaCO₃ < Gluconate.
- Gluconate is effective in the concentration range 2 - 10 mmol L⁻¹.

Bauxite residue flocculation experiments

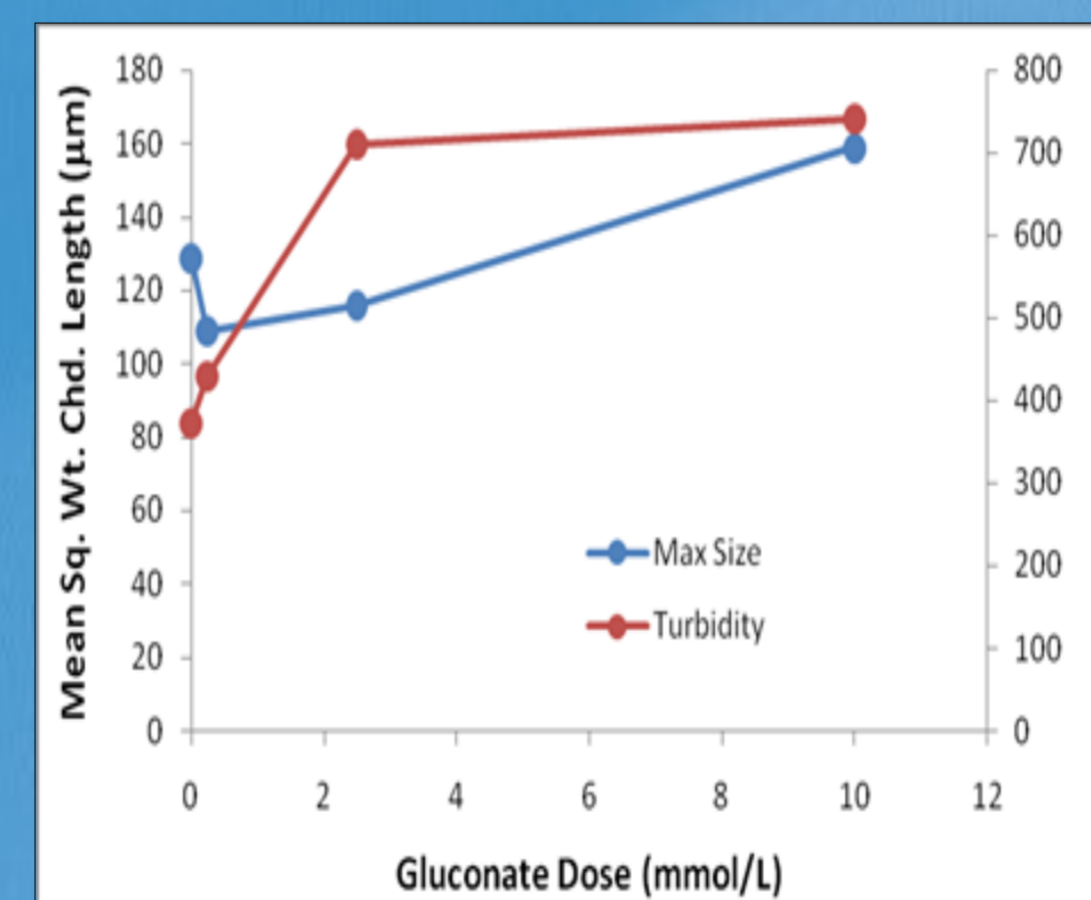


Figure 13. FBRM and turbidity data as a function of gluconate dose for the flocculation of bauxite residue with 202 g/t 9779 flocculant.

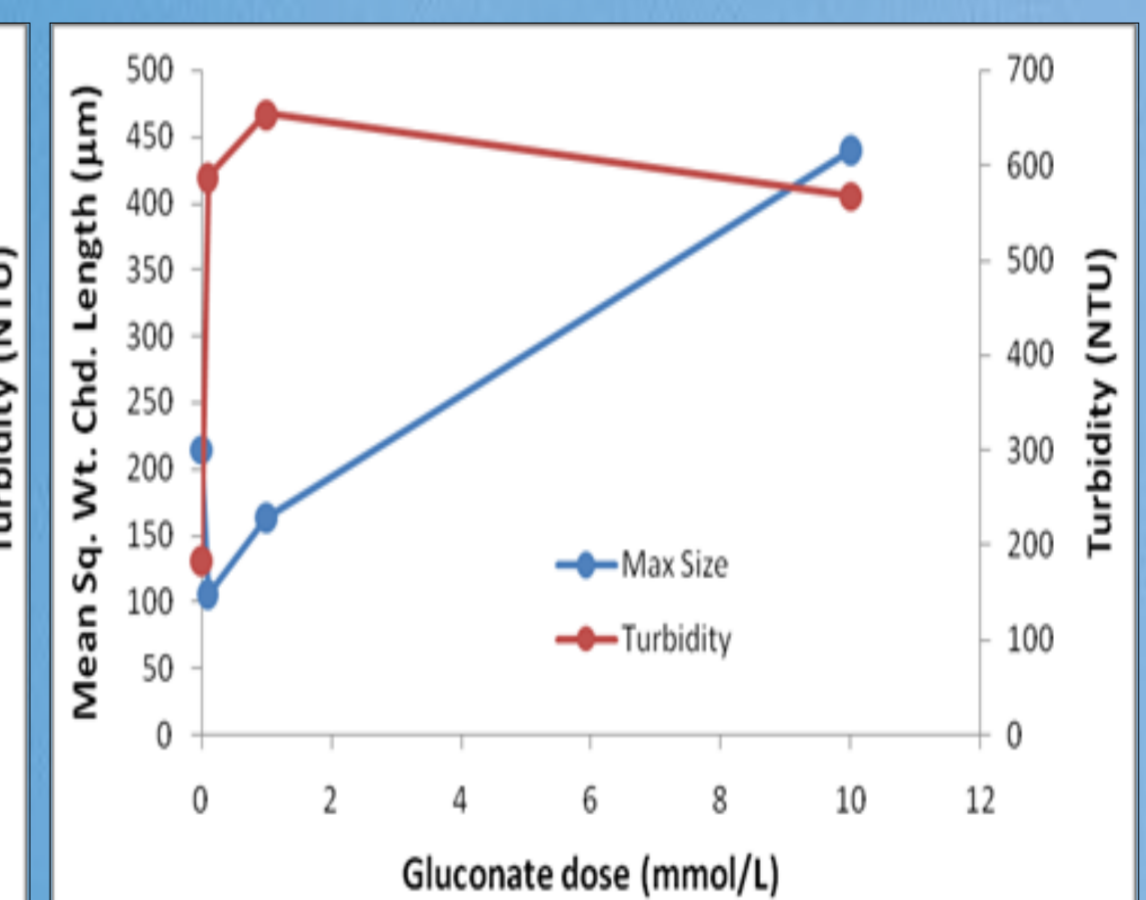


Figure 14. FBRM and turbidity data as a function of gluconate dose for the flocculation of bauxite residue with 702 g/t HX300 flocculant.

- Gluconate behaviour is complex.
- Flocculation negatively affected at low doses, but positive affected at high doses.
- The effect may involve surface adsorption or complexation of the residue/flocculant.

Gibbsite crystallisation experiments

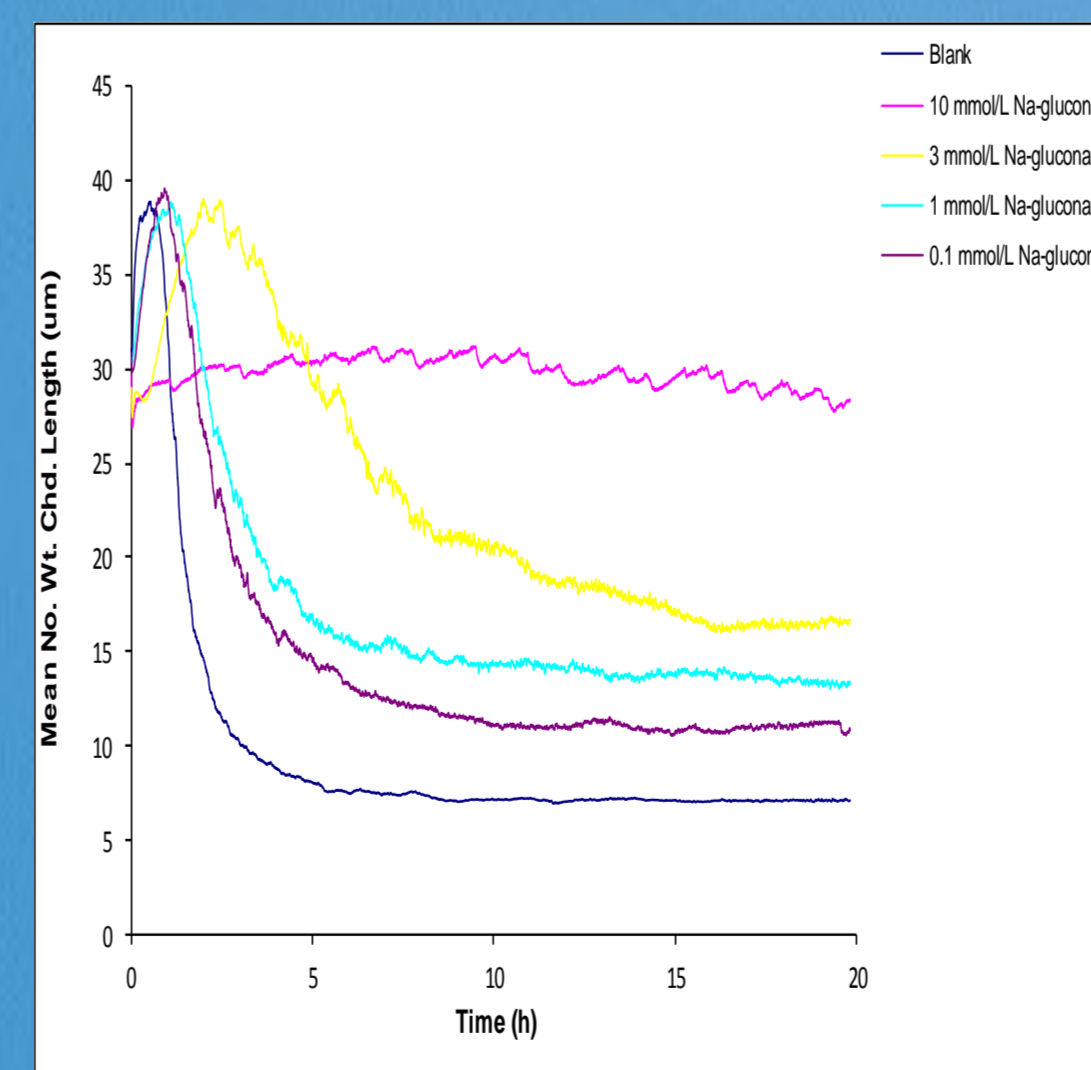


Figure 15. FBRM data showing the effect of gluconate additive on gibbsite crystallisation.

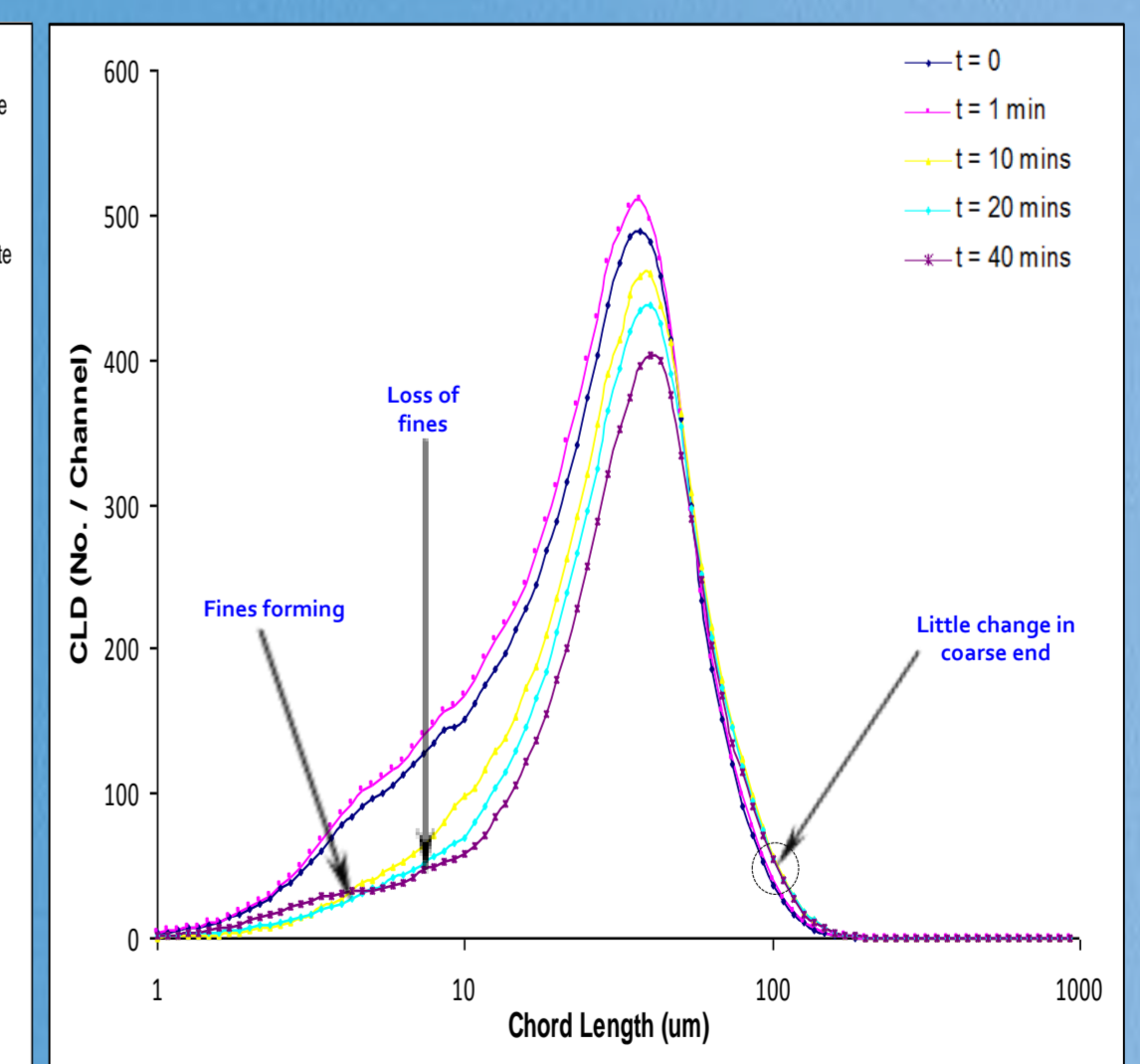


Figure 16. Change in CLD when monitoring gluconate effect on gibbsite crystallisation.

- Gibbsite crystallisation deteriorates markedly as the gluconate dose increases.
- Gluconate suppression mechanism is likely due to adsorption onto gibbsite seed surfaces.

Conclusions

- Goethite is an active seed in the premature crystallisation (auto-precipitation) of gibbsite in the mud circuit.
- Sodium gluconate is an effective additive for mitigating against premature crystallisation of gibbsite.
- Sodium gluconate negatively affects bauxite residue flocculation and gibbsite crystallisation, this warrants further investigation, due to the potential impact it may have on these unit operations.

Acknowledgements

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